

Supplementary material: Parameters derived for halogenated compounds

TABLE 1: Force field parameters for halocompounds derivatives

Bonds	K_b (kcal mol ⁻¹ Å ⁻²) ^a	r_0 (Å)
CT-CL	367.0	1.791 ^b
CT-BR	367.0	1.966 ^b
CT-XC	379.1 ^c	1.465
XC-XN	754.0 ^c	1.134

^aAs bond force constant for CT-F.

^bAveraged from different bromo- or chlorocompounds (Gaussian optimization).

^cCT-XC and XC-XN force constants calculated through interpolation.

Angles	K_θ (Kcal mol ⁻¹ rad ⁻²) ^d	θ_0 (deg) ^e
CL-CT-H1	38.1	107.7
BR-CT-H1	38.1	106.0
CT-CT-BR	50.0	110.3
BR-CT-BR	77.7	113.1
BR-CT-H2	38.1	107.1
CT-CT-CL	50.0	110.3
H1-CT-XC	50.0	107.9
CT-XC-XN	140.0	179.5

^dAssumed to be the same as bond angle constant in chloro analogues from Fox, T.; Thomas, B.E., IV; McCarrick, M.; Kollman, P.A. *J. Phys. Chem.* **1996**, *100*, 10779-10783. For acetonitrile as H1-CT-CT bond angle constant.

^eAveraged from different bromocompounds (Gaussian optimization).

Dihedrals ^f	IDIVF	K_ϕ (kcal mol ⁻¹ rad ⁻¹)	phase (deg)	periodicity
CL-CT-CT-CL	1	0.65	0.0	-3.
CL-CT-CT-CL	1	0.68	0.0	1.
X-CT-XC-X	3	0.00	0.0	1.

^fThis work.

Nonbonded parameters ^g	R^* (Å)	ϵ (kcal mol ⁻¹)
CL	1.968	0.285
BR	2.0716	0.426
XN ^h	1.740	0.210

^gThis work. ^hCyano nitrogen.